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(54) Title: NON-FOAMING CLEANING COMPOSITIONS AND A METHOD FOR THEIR USE

(57) Abstract: The present invention provides stable, non-foaming cleaning compositions having a solvent system and a surfactant system. Preferably, one composition also has a propellant system enabling the composition to be delivered as a non-foaming aerosol spray. The compositions of the present invention may also have one or more enhancing agents.



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NON-FOAMING CLEANING COMPOSITIONS AND A METHOD FOR THEIR USE

5 BACKGROUND OF THE INVENTION

I. Field of the Invention

The present invention relates to cleaning compositions. More particularly, the present invention relates to aqueous, non-foaming cleaning
10 compositions capable of removing stains from carpets, fabrics and other inanimate surfaces.

II. Description of the Prior Art

Fabric and carpet fibers may stain upon contact with water and
15 proteinaceous materials, particulate matter, oily/greasy type soils, oxidizable materials, and numerous other materials. Such stains are conventionally removed by compositions containing combinations of cleansing surfactants that lift and remove these stains from the fabric. Stain removing compositions may also be formulated to further contain an array of enhancing agents
20 including active oxygen-containing compounds, malodor reducing compounds, fragrances, and anti-soiling agents, commonly referred to as soil resists, that prevent or impede the re-soiling of the fabric after cleaning. One type of soil resist, a fluorosurfactant, is described in U.S. Patent No. 5,712,240 to Tyerech et al. See also U.S. Patent No. 5,728,669 to Tyerech. Compositions for
25 reducing malodor impressions are described in U.S. Patent No. 5,783,544 to Trinh et al. These include perfumes, metal salts and cyclodextrins.

In addition to providing acceptable stain removal ability, stain removal compositions must be stable for long periods of time under different
30 conditions, and must be convenient and safe to use. Typically, a cleaning composition is delivered to a stain by way of a substrate, such as, a wipe, a

cloth, or a sponge, or it is applied directly to the stain from the container in which it is housed. These containers may be adapted to spray or squirt the cleaning composition onto the stain and may also be fitted with an additional mechanical cleaning means, such as a brush or other textured surface, to aid
5 in removing the stain.

Sprayable cleaners have been used for many years for both household and institutional cleaning of a variety of organic and inorganic soils on all types of surfaces, including textiles, hard surfaces, and other inanimate surfaces.
10 Many of these sprayable cleaners are formulated as foaming cleaners due to the belief that the foaming action helps "lift" stains off the object or surface to be cleaned.

A number of attempts to provide fabric cleaning compositions have
15 been made. U.S. Patent Nos. 3,723,323 and 3,723,358 to Morgan et al. each describe aqueous fabric treating shampoo compositions containing anionic or nonionic surfactants as cleansing agents and neutralized polymers of acrylic or methacrylic acid with styrene or other unsaturated monomers such as alkyl acrylates and methacrylates. The polymer is said to impart anti-soiling
20 properties to fabrics cleaned with the compositions.

U.S. Patent No. 4,013,595 to Podella et al. provides non-flammable aqueous aerosol rug cleaners using hydrocarbon propellants. They possess reduced flammability due to the presence of at least 0.3% lauryl alcohol in
25 combination with 0.3% to 10% of an alkali metal lauryl sulfate salt as at least one of the surfactants. The Podella et al. compositions also contain polymers of the type taught in the Morgan et al. patents noted above.

U.S. Patent No. 3,734,686 to Douglas provides a carpet cleaning
30 shampoo composition for carpets and pile fabrics, which claims to enhance the abrasion resistance and antistatic properties of the treated carpet or pile

5 fabric. This benefit comes from the presence of an aqueous emulsion of oxygen-free polyethylene of particle size 0.02 to 0.5 microns and average molecular weight of 7,000 to 40,000 where at least 30% of the particles are covered with an emulsifier composition. This patent states that the composition possesses increased foaming action and that less force is needed to apply the shampoo to the carpet.

10 U.S. Patent No. 4,304,610 to Weisensel provides a carpet cleaning method for use with extraction machinery to clean carpets that contain high foaming anionic surfactants. The aqueous liquid or dry powder composition contains a cationic surfactant that reacts with and suppresses foaming of the anionic surfactant in the carpet, a nonionic surfactant as a primary cleaning agent, builders, fillers and chelating agents, and optionally, optical brighteners, dyes and perfumes.

15 U.S. Patent No. 4,561,992 to Troger et al. provides an aerosol cleaning agent for textile surfaces that contains plasticized urea-formaldehyde resin foam particles (0.005-0.120 mm), propellant, anti-settling agent, suspending agent, a liquid, and sodium aluminum silicate particles. A silicone defoamer
20 may be included to promote the removal of soap residue. The product is applied to textile upholstery, allowed to dry and vacuumed away from the fabric.

25 U.S. Patent No. 4,678,595 to Malik et al. provides a carpet shampoo or upholstery cleaning composition containing a glycoside surfactant, a normally solid, water-soluble or water dispersible polymer component and water. Optionally, antistatic agents, foam builders and stabilizers, such as, amine oxides and amphoteric cycloimidines or imidazolines, optical brighteners, perfumes and the like can also be included.

30

U.S. Patent No. 4,780,100 to Moll provides a foaming aqueous aerosol fabric cleaning composition, which has foaming surfactants, solvents, propellants, builders and water. The foam arguably enters the fabric pile and brings dirt up to the surface as a second foam is formed when the solvent
5 evaporates. The only requirement for surfactants is that they form a foam and can include organosilicones. Amphoteric surfactants such as betaines can be used.

The following further represent the state of the art. Additional carpet,
10 textile, and fabric cleaning compositions are provided in U.S. Patent Nos. 5,514,302 to Brown; 5,922,665 to Liu; 5,948,743 to Fonsny et al.; and 6,048,368 to Tcheou et al.

While sprayable foaming cleaning compositions are well known in the
15 art, they are not without disadvantage. Foams may contain at least 15% to about 30% air, which can result in an incomplete product to stain surface contact and subsequent cleaning ineffectiveness. Moreover, foam, when not completely removed from the surface being cleaned, can form a residue, which is not only visually undesirable, but requires additional cleaning to
20 remove the residue.

Moreover, due to environmental air quality concerns, strict laws regulating the use of chemicals, such as solvents in cleaning solutions, that emit volatile organic compounds (VOC's), are in place. As a result, many
25 current cleaning compositions with these solvents will require either discontinuance or drastic reformulation to meet the stringent VOC regulations. Without the use of these well known and effective solvents, current cleaning compositions may be ineffective.

30 To overcome these and other disadvantages, the cleaning composition of the present invention is a stable, non-foaming aerosol composition, capable

of removing stains on carpets, fabrics, and other inanimate objects. In addition, the compositions of the present invention are uniquely formulated to not only have superior stain removal properties, but to also comply with environmental regulations directed to VOC's.

5

SUMMARY OF THE INVENTION

It is an object of the present invention to provide cleaning compositions capable of removing stains from carpets, fabrics, and other inanimate objects.

10 It is another object of the present invention to provide such cleaning compositions that are non-foaming upon application to the surface to be cleaned.

It is still another object of the present invention to provide such cleaning
15 compositions that are delivered as a non-foaming aerosol spray.

It is yet another object of the present invention to provide such cleaning compositions that comply with environmental regulations directed to volatile organic compounds.

20

It is a further object of the present invention to provide a method for cleaning a stain from a carpet, fabric, or other inanimate object using the non-foaming cleaning compositions of the present invention.

25 To accomplish the foregoing objectives and advantages, the present invention, in brief summary, is a stable, non-foaming cleaning composition comprising: a solvent system and a surfactant system. In one embodiment of the present invention, the composition further includes a propellant system enabling the composition to be delivered as a non-foaming aerosol spray.

30 The present invention may also include one or more enhancing agents. The

present compositions are not only effective in removing stains, they are compliant with environmental regulations for VOC's.

5 DETAILED DESCRIPTION OF THE INVENTION

The present invention provides stable, aqueous, non-foaming cleaning compositions. Preferably, a cleaning composition of the present invention has a solvent system and a surfactant system. Still further, the cleaning composition may have one or more enhancing agents. Such enhancing agents include, but are not limited to, one or more preservatives, fragrances, malodor reducing systems, pH control agents, peroxygen components, soil resists, or any combinations thereof.

The aqueous cleaning composition of the present invention has a solvent system that aids in the removal of stains. Preferably, the solvent system includes one or more organic solvents that are not only effective at removing stains, but are present in an amount that complies with environmental regulations directed to VOC's.

Suitable organic solvents include, but are not limited to, one or more halogenated hydrocarbons, hydrocarbons, glycols, ethylene glycols, glycol ethers, propylene glycols, propylene glycol methyl ethers, propylene glycol butyl ethers, dipropylene glycols, dipropylene glycol methyl ethers, tripropylene glycols, tripropylene glycol methyl ethers, diethylene glycol butyl ethers, methanols, ethanol, propanols, isopropanols, n-butanols, sec-butanols, tert-butanols, hexylcellosolves, butylcellosolves, methylcellosolves, aromatic hydrocarbons, esters, glycol ether esters, ketones, plasticizers, or any combinations thereof.

Preferably, the solvent system includes one or more of the following organic solvents: propylene glycol monomethyl ether, propylene glycol t-butyl

ether, diethylene glycol monobutyl ether, isopropanol, or any combinations thereof.

5 The solvent system is present in the cleaning composition in an amount about 2 percent by weight (wt.%) to about 60 wt.% of the total weight of the composition. Preferably, the solvent system is present in an amount about 10 wt.% to about 50 wt.%, and more preferably about 15 wt.% to about 45 wt.%, of the total weight of the aerosol cleaning composition.

10 The cleaning composition of the present invention also has a surfactant system. The surfactant system has one or more surfactants in an amount sufficient to remove stains, yet avoid foaming. These surfactants are preferably mild surfactants. Such surfactants may be anionic, cationic, nonionic, zwitterionic, amphoteric, or any combinations thereof.

15 Preferably, the surfactant system of the present invention has one or more nonionic surfactants, amphoteric surfactants, or any combinations thereof.

20 Suitable nonionic surfactants that can be used in the cleaning compositions of the present invention include, but are not limited to, one or more ethoxylated (EO), propoxylated (PO), inter-dispersed ethoxylated-propoxylated (EO-PO), alcohols, copolymers, fatty acids, alkyl phenols, polyglycosides, n-alkylpyrrolidones, block copolymers, or any combinations
25 thereof. The most preferred nonionic surfactants are EO, PO, and EO-PO alkyl phenols, or any combinations thereof. Most preferably, the nonionic surfactant is nonyl phenol ethoxylate.

30 Suitable amphoteric surfactants include, for example, one or more betaines, amine oxides, alkyl imidazolines, cocaamphopropionates, or any combinations thereof. Preferably, the amphoteric surfactant is an

alkylhydroxypropylsultane sold under the tradename BURCO[®] HCS-989-DF by Burlington Chemical Company, Inc.

5 The surfactant system is present in the cleaning composition in an amount about 0.001 wt.% to about 2 wt.% of the total weight of the cleaning composition. Preferably, the surfactant system is present in an amount about 0.005 wt.% to about 1 wt.%, and more preferably about 0.01 wt.% to about 0.5 wt.%, of the total weight of the cleaning composition.

10 The cleaning compositions of the present invention are aqueous compositions. Preferably, the compositions include water, which is present in an amount about 35 wt.% to about 95 wt.% of the total weight of the composition. More preferably, the water is present in an amount about 40 wt.% to about 75 wt.% of the total weight of the composition.

15 Preferably, the cleaning compositions of the present invention also include one or more enhancing agents. These enhancing agents improve the stability, efficacy, ease, and/or pleasure of use by the end user. The one or more enhancing agents that can be used in the present cleaning compositions
20 include, but are not limited to, one or more preservatives, fragrances, malodor reducing agents, salts, pH control agents, peroxygen components, soil resists, or any combinations thereof.

25 Suitable preservatives include, but are not limited to, one or more organic salts, organic sulfur compounds, halogenated compounds, cyclic organic nitrogen compounds, low molecular weight aldehydes, quaternary compounds, phenyl and phenoxy compounds, parabens, organic acids and derivatives, iodophors, quaternary ammonium compounds, urea derivatives, isothiazolines, alkyl substituted amino acids, formaldehydes, formaldehyde
30 donors including 1-(3-chloroallyl)-3,5,7-azoniaadamantane chloride, 1,3-bis(hydroxymethyl)-5,5-dimethylhydantoin, 1-(hydroxymethyl)-5,5-dimethyl

hydantoin, 3-iodo-2-propynyl butyl carbamate, hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine, or any combinations thereof. Preferably, the preservative is an organic salt, such as, for example, sodium benzoate.

5 When a preservative is present, it is present in an amount about 0.001 wt.% to about 2 wt.% based on the total weight of the composition. Preferably, the preservative is present at about 0.01 wt.% to about 1 wt.%, with about 0.05 wt.% to about 0.5 wt.% being the most preferred amount.

10 One or more fragrances may be used in the present cleaning compositions to eliminate the chemical smell inherent with cleaning compositions. Also, the fragrance may mask or eliminate any odors associated with the stain being cleaned.

15 Suitable fragrances include, but are not limited to, one or more aromatic or aliphatic esters having a molecular weight from about 130 to about 250, aliphatic or aromatic alcohol having a molecular weight from about 90 to about 240, aliphatic ketone having a molecular weight from about 150 to about 260, aromatic ketone having a molecular weight from about 150 to
20 about 270, aromatic and/or aliphatic lactone having a molecular weight from about 130 to about 290, aliphatic aldehyde having a molecular weight from about 140 to about 200, aromatic aldehyde having a molecular weight from about 90 to about 230, aliphatic or aromatic ether having a molecular weight from about 150 to about 270, condensation product of aldehyde and amine
25 having a molecular weight from about 180 to about 320, or any combinations thereof.

 When present, the one or more fragrances are present in an amount about 0.001 wt.% to about 2 wt.% of the total weight of the composition.

30 Preferably, the fragrance is present in an amount about 0.05 wt.% to about 1

wt.%. The more preferred amount of fragrance is about 0.1 wt.% to about 0.5 wt.% of the total weight of the composition.

5 One or more malodor reducing agents may be used in the cleaning composition of the present invention. Useful malodor reducing agents include, but are not limited to, one or more cyclodextrins, inorganic and organic acids, metallic salts of inorganic and organic acids, enzymes or enzyme systems, metazenes, or any combinations thereof.

10 One or more peroxygen compounds may be used in the present invention to enhance cleaning properties of the compositions. Suitable peroxygen compounds include, but are not limited to, hydrogen peroxides, t-butyl hydroperoxides, sodium or other salts of percarbonates, sodium or other salts of perborates, or any combinations thereof.

15 One or more pH control agents may be used in the present invention. Suitable pH control agents include, but are not limited to, one or more inorganic acids, inorganic bases, salts of inorganic acids, organic acids, organic bases, salts of organic acids, phosphorous compounds, or any combinations thereof.

20

The pH range of the cleaning composition is about 4 to about 12. The more preferred pH range is about 6 to about 10, with the most preferred being about 6.5 to about 9.

25 The cleaning composition of the present invention may also have one or more soil resists. The soil resist may be, for example, polymeric or copolymeric. Suitable polymeric or copolymeric soil resists include, but are not limited to, one or more polymers derived from styrene-maleic anhydride copolymer resins, monomers of acrylic acids, methacrylic acids,

30 methacrylates, methylmethacrylates, ethyl acrylates, maleic acids, copolymers

derived from the above monomers, and any combinations thereof. Still others may include, but are not limited to, one or more polyvinylpyrrolidones, polyacrylates, modified cellulose polymers, polycarboxylates, vinyl acetate/maleic anhydride copolymer resins, cationic amines, aliphatic quaternary ammonium salts known to have anti-static properties, imidazoline salts, fluoroaliphatic oligomer or polymers, fluorinated hydrocarbon soil resists such as the Zonyl[®] series from DuPont, fluorinated acrylate copolymers, or any combinations thereof.

10 As described above, the cleaning compositions are formulated to be non-foaming when dispensed and/or applied to a stain. As such, the cleaning compositions of the present invention may be contained in any suitable packaging or containers known to those in the art. These packages or containers may include, for example, bottles, such as spray or squirt bottles, aerosol canisters, substrates such as wipes or sponges, and/or barrier-type packages.

In a preferred embodiment of the present invention, the non-foaming cleaning compositions are packaged in an aerosol canister and are capable of being delivered to a stained article as a non-foaming spray. When packaged as an aerosol, the compositions of the present invention include a propellant system. The propellant system may include, for example, propellants, such as, hydrocarbons and/or inorganic gases. Suitable hydrocarbon propellants include, for example, methane, ethane, n-propane, n-butane, isobutane, n-pentane, isopentane, or any mixtures thereof. Suitable inorganic gas propellants include, for example, carbon dioxide, nitrogen, nitrous oxide, or any combinations thereof.

When the cleaning compositions of the present invention are formulated as aerosol compositions, the propellant system is present in an amount about 5 wt.% to about 10 wt.% of the total weight of the aerosol

composition. Preferably, the propellant system is present in an amount about 7 wt.% to about 9 wt.% of the total weight of the aerosol composition.

To remove a stain on a carpet, fabric, or other inanimate object, the cleaning composition of the present invention is first applied to the stain. The method by which the cleaning composition is applied will vary with the type of package or container that houses the cleaning composition. Once the cleaning composition is applied to the stain, the consumer uses a substrate like a towel, cloth, and/or sponge to blot, dab, and/or rub the stained area to remove the stain. While it is apparent from the test data set forth below that the compositions of the present invention are effective at removing stains after one or two applications, the cleaning procedure may be repeated until the consumer is satisfied with the level of stain removal.

Non-foaming cleaning compositions of the present invention were formulated with the following chemical constituents in amounts expressed as percent by weight of the total weight of cleaning composition.

Option #1 Cleaning Composition

Chemical Name	Trade Name	Concentrate %	Aerosol %
DI Water	DI Water	71.45	65.74
Sodium Benzoate	Sodium Benzoate	0.2	0.18
Amphoteric Surfactant	Burco HCS-989-DF*	0.2	0.18
Propylene Glycol Monomethyl Ether	Arcosolv PM/Dowanol PM	10	9.2
Propylene Glycol t-butyl Ether	Arcosolv PTB	8	7.36
Diethylene Glycol	Butyl Carbitol / Dowanol D	10	9.2

Monobutyl Ether			
Fragrance	Fragrance	0.15	0.14
Hydrocarbon Propellant A-46	A-46	-----	8
TOTAL		100	100

*Proprietary material (55% active) (50% Trade Secret, 5% Proprietary Surfactant)

Option #2 Cleaning Composition

5

Chemical Name	Trade Name	Concentrate %	Aerosol %
DI Water	DI Water	51.73	47.6
Sodium Benzoate	Sodium Benzoate	0.33	0.3
Nonyl Phenol Ethoxylate	Igepal CO-630 / Cedepa CO-630	0.33	0.3
Isopropyl Alcohol	Isopropyl Alcohol	18.15	16.7
Diethylene Glycol Monobutyl Ether	Butyl Carbitol / Dowanol DB	29.13	26.8
Fragrance	Fragrance	0.33	0.3
Hydrocarbon Propellant A-46	A-46	-----	8
TOTAL		100	100

The following examples of stain removal performance illustrate the effectiveness of the cleaning compositions of the present invention when used to remove stains from carpets and fabrics.

10

15

Example 1: Product Consumer Testing

The performance and consumer response of Option 1 cleaning composition and Option 2 cleaning composition were evaluated as new products for removing spots and stains from carpets and fabrics. The product testing group employed a total of 200 people from their database to participate in these studies. Women in the studies were people who clean spots and stains from their carpets, and who have done so within the last 3 months. Two monadic tests were performed, one for each product. One hundred panelists were in each test. Each product was positioned as an aerosol Carpet Spot & Stain Remover. Both formulations were placed in white 12 oz. aluminum cans equipped with an Accusol Trigger. Each product was used for a total of two weeks by each panelist in the test. The results of the tests are set forth in Tables 1 through 3 below.

Table 1. Consumer Testing Results for Option 1 and Option 2 Cleaning Compositions

	Option #1*	Option #2*
	Top Box/Top 2 Box	Top Box/Top 2 Box
Cleaning		
Cleans well overall	78 / 89	62 / 84
Effectively removes spots and stains	72 / 90	59 / 82
Removes old stains	49 / 66**	34 / 53**
Removes fresh stains	77 / 87**	68 / 86**
Has a powerful, deep penetrating spray	76 / 97	60 / 87
Does not leave a residue	82 / 98	68 / 90
Convenience		
Is easy to use	88 / 98	70 / 89

Solves a problem for you	68 / 90**	62 / 83
Is convenient to use	90 / 95	68 / 89
Is easy to handle	87 / 96	69 / 89
Has an easy to direct spray	85 / 97	82 / 94
Fragrance/Odors		
Leaves carpet smelling clean and fresh	54 / 71	62 / 85**
Has a pleasant scent	49 / 66	55 / 78
Safety		
Is safe for your carpets	80 / 98	66 / 86**
Does not discolor carpet	90 / 99	82 / 94

*(Number of Respondents – Top Box = Strongly Agree / Top 2 Box = Strongly Agree + Somewhat Agree)

** (Number of respondents is less than 100)

5

Table 2: Cleaning Performance (%)

	Option #1	Option #2
Complete Removal of Stains	79	71
Applications Needed For Effective Cleaning		
Two or less	90	91
More than two	10	9
Comparison to Current Product		
Better than	50	52
Equal to	39	32

10

Table 3: Physical Attributes of Compositions (%)

	Option #1	Option #2
Amount of Foam		
No Foam	42	30
Just Right	58	62
Too Much Foam	0	8

- 5 Overall each embodiment of the present invention was well received by participants of the studies. High acceptance ratings on all attributes demonstrate how well the compositions of the present invention performed.

EXAMPLE 2: Product Efficacy Testing

Testing was performed in the laboratory to demonstrate the ability of the compositions of the present invention in removing common stains typically found on rugs and carpets. Cut pile 100% nylon carpet swatches (4" x 4" in dimension) were used for test purposes. Liquid and semi-viscous liquids were uniformly applied consistent with standard laboratory practice. The cleaning procedure was initiated after a one hour dwell period. A second study was done where the cleaning procedure was initiated after a 24 hour dwell period. The test procedure used was as follows:

10

1. The product aerosol can was vigorously shaken, then product(equivalent to 6.5 grams of product) was sprayed onto the stain for 1 second.
2. The stain area was blotted with a clean, white paper towel until no
- 15 visible signs of stain transfer were apparent.
3. Steps 1 and 2 were repeated a second time.

4. Treated carpet swatches were allowed to dry at ambient temperature overnight, then vacuumed using five forward and five reverse strokes with a commercial upright beater bar vacuum.

5. An internal panel of ten people assessed the appearance of the stain using the AATCC Gray Scale (see below).

Gray Scale Description

10	5	Negligible or no staining
	4	Slightly stained
	3	Noticeably stained
	2	Considerably stained
	1	Heavily stained

15

The scale is used to compare a stain that has been cleaned against two controls: an unstained carpet swatch and a stained carpet swatch. A rating of 5 indicates that the remnants of the stain are negligible or undetectable, and a rating of 1 indicates a severe difference between the stain that has been cleaned against an unstained carpet swatch.

20

Results of the one hour and 24 hour studies are set forth in Table 4A and Table 4B, respectively.

25 Table 4A: Soil Removal Results – One Hour Dwell Period

STAIN	Option #1	Option #2
Coffee	4.2	3.9
Cola	5.0	4.8
Grape Juice	4.9	4.9
Shoe Polish	4.2	3.7

Red Wine	4.8	4.9
----------	-----	-----

Table 4B: Soil Removal Results – 24 Hour Dwell Period

5

STAIN	Option #1	Option #2
Coffee	3.8	3.6
Cola	5.0	5.0
Grape Juice	5.0	4.8
Shoe Polish	3.8	2.2
Red Wine	4.9	5.0

Results of these two studies clearly indicate the effectiveness of the cleaning compositions of the present invention in removing stains from carpet.

10

A second cleaning study was performed using Option #1 cleaning composition and Option #2 cleaning composition where mud and dirty motor oil were used as the stains. The testing procedure was similar to that described above except that only a 24 hour dwell period was used. The stains were each cleaned using one application of product (7 grams), allowed to dry overnight, and then rated by a 10 person panel test. Once rated, the soiled carpet swatches were then cleaned a second time using an additional application of product (7 grams), allowed to dry overnight, vacuumed, then rated using another 10 person panel test. Results of this study are shown in Table 5.

20

Table 5. Soil Removal Results – After One and Two Product Applications

		One Application		Two Applications		
		Mud	Dirty Motor Oil		Mud	Dirty Motor Oil
Option #1		3.3	3.4		5.0	4.6
Option #2		4.3	3.7		4.9	4.2

5

Results of this study clearly indicate the effectiveness of the cleaning compositions of the present invention at removing tough common stains typically found on rugs and carpets.

10 A third cleaning study was performed using Option #1 cleaning composition and Option #2 cleaning composition. For this example of cleaning performance, testing was performed by an independent testing laboratory to demonstrate the ability of both Option #1 and Option #2 cleaning compositions to remove common stains typically found on rugs and carpets.

15 Thirteen stains were evaluated. Test material used was continuous filament, cut pile, beige color stainmaster treated carpet, 100% nylon. Liquid and semi-viscous liquids were uniformly applied in a one inch diameter ring. The balances of the staining agents were applied as uniformly as possible, consistent with standard laboratory practice. All stains were applied in
20 duplicate. The cleaning procedure was initiated after a 24 hour dwell period. The cleaning procedure employed was as follows:

1. The stain was blotted with paper towels until no more stain could be absorbed.

25 2. The aerosol can containing the product to be tested was vigorously shaken for 2-3 seconds, then 7 grams of the aerosol liquid were applied to the stain.

3. The sprayed area was allowed to sit for approximately 1 minute.
4. Using a clean, white absorbent cloth or plain white paper towel, the product was blotted into the stain until no visible signs of stain transfer were apparent.
- 5 5. The cleaned area was gently wiped with clean, cool water using a clean, white, colorfast sponge, cloth or paper towel. The area was then blotted dry.
6. The test material was allowed to dry at ambient temperature and then vacuumed.
- 10 7. A panel of three technicians assessed the appearance of the stain using the AATCC Gray Scale (see earlier description).
8. Staining agents not removed to a rating of 5 were cleaned a second time. (Steps 2 – 7 were repeated).
- 15 Results of this study are set forth in Table 6.

TABLE 6 (A): Soil Removal 1st Cleaning Cycle Results – Outside Testing Facility

	<u>Staining Agent</u>	<u>Option #1</u>	<u>Option #2</u>
	Mud	5.0	4.5
	Foot Traffic Dirt	4.8	4.5
20	Coffee (Milk/Sugar)	4.5	4.0
	Dog Urine	5.0	5.0
	Red Wine	4.3	5.0
	Lipstick	3.5	3.8
	Spaghetti Sauce	4.3	3.5
25	Motor Oil – Used	4.3	4.6
	Grease – Used	4.3	3.5
	Ink – Black	4.3	4.5
	Fruit Juice	4.8	5.0

Grape Juice	4.8	5.0
Shoe Polish – Black	3.8	3.6

TABLE 6 (B): Soil Removal 2nd Cleaning Cycle Results – Outside Testing Facility

	<u>Staining Agent</u>	<u>Option #1</u>	<u>Option #2</u>
5	Mud	5.0	5.0
	Foot Traffic Dirt	4.9	5.0
	Coffee (Milk/Sugar)	4.5	4.5
	Dog Urine	5.0	5.0
	Red Wine	4.8	5.0
10	Lipstick	3.8	4.0
	Spaghetti Sauce	4.5	4.5
	Motor Oil – Used	4.8	4.7
	Grease – Used	5.0	4.5
	Ink – Black	4.5	5.0
15	Fruit Juice	5.0	5.0
	Grape Juice	5.0	5.0
	Shoe Polish – Black	4.5	4.8

Results from this study, as well as the first two cleaning studies, clearly demonstrate the effectiveness of these cleaning compositions at removing a variety of tough common stains that one could encounter on rugs, carpets and other areas and objects.

25 EXAMPLE 3: Product Stability Testing

Samples of both Option #1 and Option #2 were placed into stability studies utilizing 12 oz. aluminum cans equipped with an Accusol Trigger. The

can stability results after one month at room temperature, 100°F, 120°F, and 130°F showed no signs of corrosion under these conditions. The can stability results after two months and three months at room temperature, 100°F and 120°F, showed no signs of corrosion under these conditions.

5

For solution stability after one month at room temperature, 100°F and 120°F, comparison of the liquid concentrate of both options derived from the cans with that of concentrate samples that were kept under the same conditions but in glass jars (control samples) demonstrated no signs of discoloration, phase separation or precipitation.

10

Stability of both option formulations after three freeze/thaw cycles remained acceptable with solutions that remained clear with no visual phase separation or precipitation.

15

Overall, the above Examples not only demonstrate the effectiveness of the cleaning compositions of the present invention in removing stains from carpets and fabrics, but also the overall acceptance by consumers with respect to convenience, safety and fragrance.

20

The present invention has been described with particular reference to the preferred forms thereof. It will be obvious to one of ordinary skill in the art that changes and modifications may be made therein without departing from the spirit and scope of the present invention as defined by the following claims.

25

WHAT IS CLAIMED IS:

1. An aqueous cleaning composition comprising:
a solvent system; and
5 a surfactant system,
wherein the composition is non-foaming upon application to a stain.
2. The composition of claim 1, wherein said solvent system includes
one or more organic solvents.
10
3. The composition of claim 2, wherein said one or more organic
solvents are selected from the group consisting of halogenated hydrocarbon,
hydrocarbon, glycol, ethylene glycol, glycol ether, propylene glycol, propylene
glycol methyl ether, propylene glycol butyl ether, dipropylene glycol,
15 dipropylene glycol methyl ether, tripropylene glycol, tripropylene glycol methyl
ether, diethylene glycol butyl ether, methanol, ethanol, propanol, isopropanol,
n-butanol, sec-butanol, tert-butanol, hexylcellosolve, butylcellosolve,
methylcellosolve, aromatic hydrocarbon, ester, glycol ether ester, ketone,
plasticizer, and any combinations thereof.
20
4. The composition of claim 2, wherein said one or more organic
solvents are selected from the group consisting of propylene glycol
monomethyl ether, propylene glycol t-butyl ether, diethylene glycol monobutyl
ether, isopropanol, and any combinations thereof.
25
5. The composition of claim 1, wherein said solvent system is
present in an amount about 2 wt.% to about 60 wt.% of the total weight of the
composition.

6. The composition of claim 1, wherein said solvent system is present in an amount about 10 wt.% to about 50 wt.% of the total weight of the composition.

5 7. The composition of claim 1, wherein said surfactant system is at least one surfactant selected from the group consisting of an anionic surfactant, a cationic surfactant, a nonionic surfactant, a zwitterionic surfactant, an amphoteric surfactant, and any combinations thereof.

10 8. The composition of claim 1, wherein said surfactant system is at least one surfactant selected from the group consisting of a nonionic surfactant, an amphoteric surfactant, and any combinations thereof.

15 9. The composition of claim 8, wherein said one or more nonionic surfactants are selected from the group consisting of ethoxylated (EO), propoxylated (PO), inter-dispersed ethoxylated-propoxylated (EO-PO), and any combinations thereof.

20 10. The composition of claim 9, wherein said nonionic surfactant is nonyl phenol ethoxylate.

25 11. The composition of claim 8, wherein said one or more amphoteric surfactants are selected from the group consisting of betaine, amine oxide, alkyl imidazoline, cocaamphopropionate, alkylhydroxypropylsultane, and any combinations thereof.

12. The composition of claim 11, wherein said amphoteric surfactant is alkylhydroxypropylsultane.

30 13. The composition of claim 1, wherein said surfactant system is about 0.001 wt.% to about 2 wt.% of the total weight of the composition.

14. The composition of claim 1, wherein said surfactant system is about 0.005 wt.% to about 2 wt.% of the total weight of the composition.

5 15. The composition of claim 1, further comprising one or more enhancing agents.

16. The composition of claim 15, wherein said one or more enhancing agents are selected from the group consisting of: preservative,
10 fragrance, malodor reducing agent, salt, pH control agent, peroxygen component, soil resist, and any combinations thereof.

17. The composition of claim 16, wherein said preservative is selected from the group consisting of organic salt, organic sulfur compound,
15 halogenated compound, cyclic organic nitrogen compound, low molecular weight aldehyde, quaternary compound, phenyl and phenoxy compound, paraben, organic acid, organic acid derivatives, iodophor, quaternary ammonium compound, urea derivative, isothiazoline, alkyl substituted amino acid, formaldehyde, formaldehyde donors, 1-(3-chloroallyl)-3,5,7-
20 azoniaadamantane chloride, 1,3-bis(hydroxymethyl)-5,5-dimethylhydantoin, 1-(hydroxymethyl)-5,5-dimethyl hydantoin, 3-iodo-2-propynyl butyl carbamate, hexahydro-1,3,5-tris(2-hydroxyethyl)-s-triazine, and any combinations thereof.

18. The composition of claim 16, wherein said preservative is about
25 0.001 wt.% to about 2 wt.% of the total weight of the composition.

19. The composition of claim 16, wherein said fragrance is selected from the group consisting of aromatic or aliphatic ester having a molecular weight from about 130 to about 250, aliphatic or aromatic alcohol having a
30 molecular weight from about 90 to about 240, aliphatic ketone having a molecular weight from about 150 to about 260, aromatic ketone having a

molecular weight from about 150 to about 270, aromatic and/or aliphatic lactone having a molecular weight from about 130 to about 290, aliphatic aldehyde having a molecular weight from about 140 to about 200, aromatic aldehyde having a molecular weight from about 90 to about 230, aliphatic or
5 aromatic ether having a molecular weight from about 150 to about 270, condensation product of aldehyde and amine having a molecular weight from about 180 to about 320, and any combinations thereof.

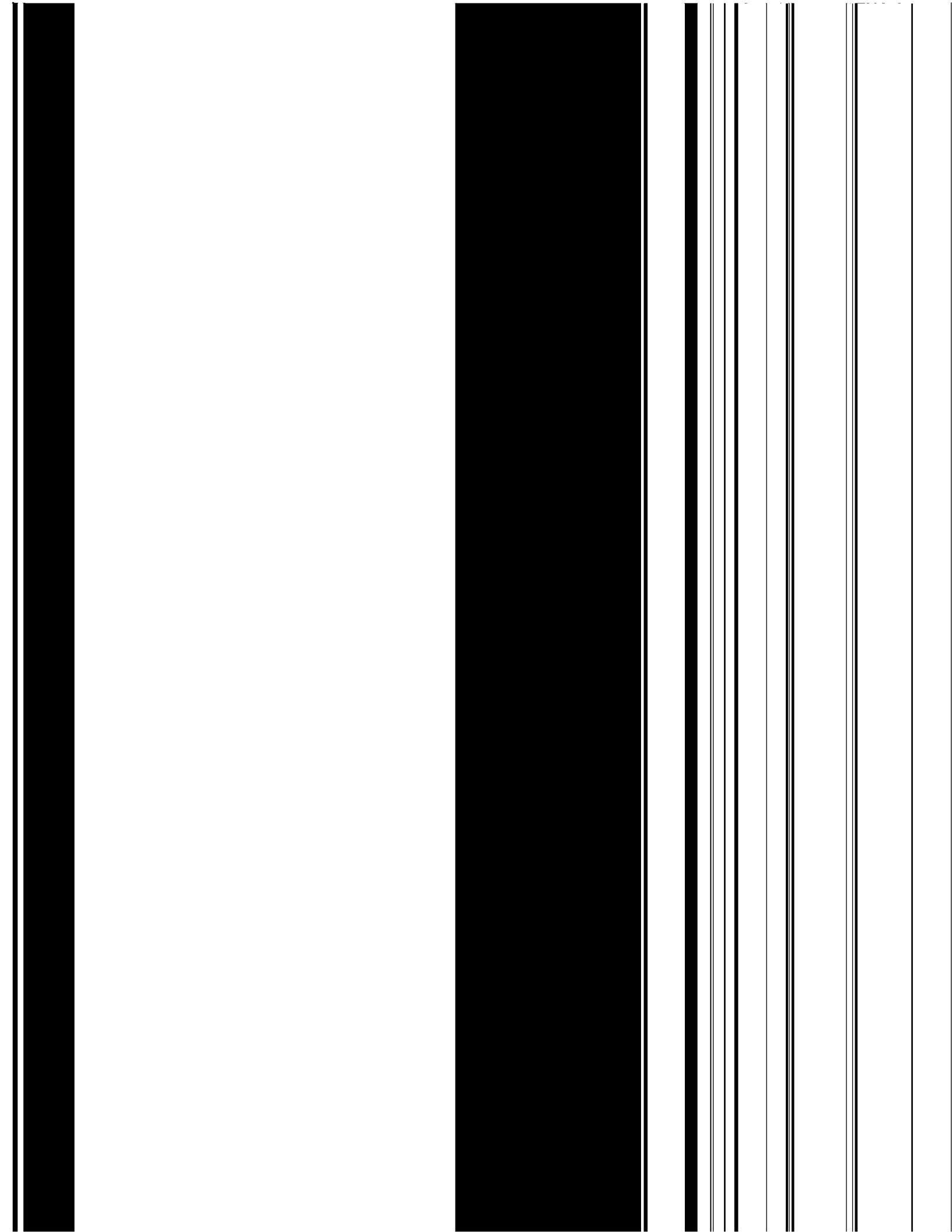
20. The composition of claim 16, wherein said fragrance is about
10 0.001 wt.% to about 2 wt.% of the total weight of the composition.

21. The composition of claim 1, further comprising about 35 wt.% to about 95 wt.% water.

15 22. An aqueous aerosol cleaning composition comprising:
a solvent system;
a surfactant system; and
a propellant system,
wherein the composition is non-foaming upon application to a stain.

20 23. The composition of claim 22, wherein said solvent system includes one or more organic solvents.

24. The composition of claim 23, wherein said one or more organic
25 solvents are selected from the group consisting of halogenated hydrocarbon, hydrocarbon, glycol, ethylene glycol, glycol ether, propylene glycol, propylene glycol methyl ether, propylene glycol butyl ether, dipropylene glycol, dipropylene glycol methyl ether, tripropylene glycol, tripropylene glycol methyl ether, diethylene glycol butyl ether, methanol, ethanol, propanol, isopropanol,
30 n-butanol, sec-butanol, tert-butanol, hexylcellosolve, butylcellosolve,



31. The composition of claim 22, wherein said surfactant system is about 0.001 wt.% to about 2 wt.% of the total weight of the composition.

32. The composition of claim 22, wherein said propellant system
5 includes one or more propellants selected from the group consisting of hydrocarbons, inorganic gases, and any combinations thereof.

33. The composition of claim 32, wherein said one or more
10 hydrocarbon propellants are selected from the group consisting of methane, ethane, n-propane, n-butane, isobutane, n-pentane, isopentane, and any combinations thereof.

34. The composition of claim 32, wherein said one or more inorganic
15 gas propellants are selected from the group consisting of carbon dioxide, nitrogen, nitrous oxide, and any combinations thereof.

35. The composition of claim 22, wherein said propellant system is about 5 wt.% to about 10 wt.% of the total weight of the composition.

20 36. The composition of claim 22, wherein said propellant system is about 7 wt.% to about 9 wt.% of the total weight of the composition.

37. The composition of claim 22, further comprising one or more
25 enhancing agents.

38. The composition of claim 37, wherein said one or more
enhancing agents are selected from the group consisting of preservative,
fragrance, malodor reducing agent, salt, pH control agent, peroxygen
component, soil resist, and any combinations thereof.

30

39. A method for removing a stain from a carpet or fabric comprising the steps of:

applying a non-foaming cleaning composition having a solvent system and a surfactant system; and

5 removing said non-foaming cleaning composition and said stain from said carpet or fabric.

40. A method for removing a stain from a carpet or fabric comprising the steps of:

10 applying a non-foaming aerosol cleaning composition having a solvent system, a surfactant system, and a propellant system; and

removing said non-foaming aerosol cleaning composition and said stain from said carpet or fabric.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US02/11816

A. CLASSIFICATION OF SUBJECT MATTER

IPC(7) :C11D 1/72, 1/88, 1/94, 3/43, 3/44

US CL :Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 510/279, 280, 302, 303, 319, 336, 337, 340, 341, 342, 350, 351, 406, 506

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched
none

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

WEST

search terms: propellant, fragrance, perfume, surfactant, propylene glycol, glycol ether

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 6,150,318 A (SILVESTER et al) 21 November 2000, See Abstract; col. 3, line 1 to col. 6, line 65; claim 1.	1-33, 35-40
X	US 6,136,770 A (CHEUNG et al) 24 October 2000, See Abstract; col. 3, line 30 to col. 7, line 40; col. 10, line 5 to col. 14, line 41.	1-17, 19-32, 34, 37, 38
Y	US 6,211,124 A (ORMEROD, IV et al) 03 April 2001, See Abstract; col. 2, line 10 to col. 8, line 35.	1-40



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"A" document defining the general state of the art which is not considered to be of particular relevance	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"E" earlier document published on or after the international filing date	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&" document member of the same patent family
"O" document referring to an oral disclosure, use, exhibition or other means	
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

14 JUNE 2002

Date of mailing of the international search report

12 JUL 2002

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INTERNATIONAL SEARCH REPORT

International application No.

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A. CLASSIFICATION OF SUBJECT MATTER:

US CL :

510/279, 280, 302, 303, 319, 336, 337, 340, 341, 342, 350, 351, 406, 506